

# Nitroimidazoles: Part XVI—Some 1-Methyl-4-nitro-5-substituted Imidazoles†‡

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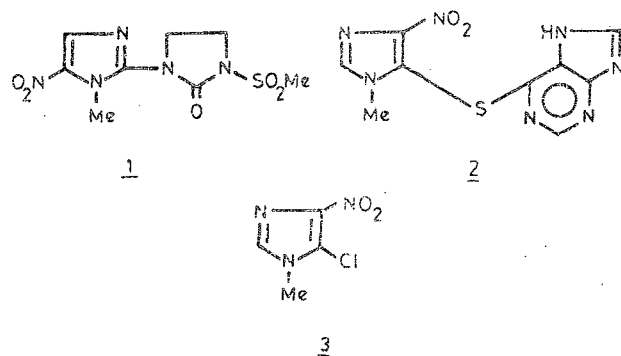
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Received 26 May 1982; accepted 22 June 1982

Treatment of 1-methyl-4-nitro-5-chloroimidazole (3) with 5-membered lactams, e.g. imidazolidinones, oxazolidinone and thiazolidinone affords N-imidazolyl derivatives (4a-d). Reaction of 3 with imidazole yields 4e; amino derivatives (4f-h) are similarly obtained. 2-Hydroxypyrazine, 4-hydroxyquinazoline and 3,4,5-trichlorophenol and 3 react to form O-derivatives (4i-k), while mercaptans provide the sulphides (4l-n). Imidazole (11) is formed from 1-methyl-4-chloro-5-nitroimidazole (10).

Our exercises in nitroimidazole chemistry resulting in the synthesis of the potent antiprotozoal agent, 1-methyl-sulphonyl-3-(1-methyl-5-nitroimidazol-2-yl)-2-imidazolidinone<sup>1</sup> (1)\*\* and the well-known immunosuppressive properties of azathioprine {[6-(1-methyl-4-nitro-imidazol-5-yl)thio]purine; imuran® 2} prompted us to synthesise some 1-methyl-4-nitroimidazoles carrying diverse substituents at position-5. These were prepared from 1-methyl-4-nitro-5-chloroimidazole (3)<sup>3</sup> by reaction with appropriate nucleophiles.

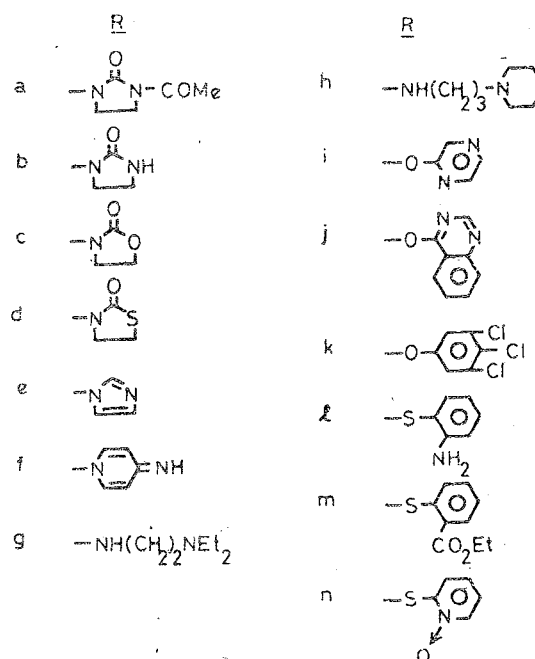
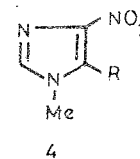
Treatment of 3 with the sodium salt of 1-acetyl-2-imidazolidinone gave a mixture of the expected 4a; the desacetyl derivative (4b) and the bis condensation product (5) of 2-imidazolidinone with 3, these were easily separated by chromatography. 5 was also obtained by reacting 2 mol of 3 with the sodium salt of



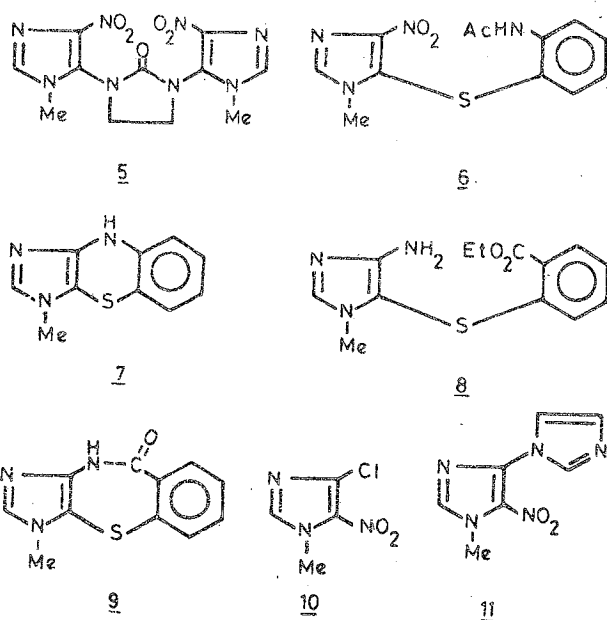
†Contribution No. 653 from CIBA-GEIGY Research Centre, Bombay 400 063.

‡Part XV: *Indian J Chem*, 21B (1982), 1078.

\*\*Code No. C16213-Go.



1 mol of 2-imidazolidinone. The reaction of 1-acetyl-2-imidazolidinone with 1-methyl-2-methanesulphonyl-5-nitroimidazole has been reported earlier to give a similar series of three products.<sup>4</sup> The products in these cases as well as with oxazolidinone and thiazolidinone are so formulated (heteroarylation of N rather than O) because of the presence of appropriate C=O bands in the IR spectra. Reaction products of the sodium salts of 2-hydroxypyrazine and 4-hydroxyquinazoline on the other hand appeared to be the O-imidazolyl derivatives 4i and 4j respectively although N-arylation is not totally ruled out. The product from 4-aminopyridine and 3 was assigned the usual nuclear N-arylated structure (4f), 4l obtained from *ortho*-aminothiophenol and 3 was converted into the acetyl derivative 6, which failed to give 7 when subjected to the usual basic conditions for the formation of phenothiazines<sup>5</sup>. Equally unfruitful was the effort to cyclise the amine (8) derived from 4m to the imidazobenzothiazepine (9). The reaction of sodiumimidazole with 3 and the isomeric halide 10<sup>3</sup> gave 4e and 11 respectively which are position isomers of the potent antiprotozoal agent, 1-(1-methyl-5-nitroimidazol-2-yl) imidazole<sup>6</sup>.



1-Acetyl-3-(1-methyl-4-nitroimidazol-5-yl)-2-imidazolidinone (4a), 1-(1-methyl-4-nitroimidazol-5-yl)-2-imidazolidinone (4b) and 1,3-bis-(1-methyl-4-nitroimidazol-5-yl)-2-imidazolidinone (5)—To a suspension of 50% NaH (3.6 g) in DMF (20 ml) was added under cooling (10–15°) and stirring, a solution of N-acetyl-ethyleneurea (9.6 g) in DMF (60 ml). After the addition was over the reaction mixture was stirred at 15–20° for 45 min. A solution of 3 (8.05 g) in DMF (30 ml) was then added and the mixture heated at 95–100° for 90 min. The solvent was removed under reduced pressure, the residue dissolved in methylene chloride, filtered and the filtrate evaporated *in vacuo*. The residue was washed with hexane to remove mineral oil and

trituated with CH<sub>3</sub>CN. The solid was filtered off and recrystallised from DMF-ether to afford 5 (2 g).

The mother liquor was evaporated *in vacuo* and the residue chromatographed over silica gel. Fractions with 2% methanolic chloroform gave N,N'-diacetyl-ethyleneurea. Further fractions with 3% methanolic chloroform gave a solid which was recrystallised from acetonitrile-ether to afford 4a. Fractions with 5% methanolic chloroform yielded 4b which was recrystallised from acetonitrile-ether. Physical properties of these and other compounds made by the above method are recorded in Table 1.

1-Methyl-4-nitro-5-(4-imino-1,4-dihydro-1-pyridyl)imidazole hydrochloride (4f)—A mixture of 3 (8.1 g) and 4-aminopyridine (4.7 g) in absolute ethanol (100 ml) was heated under reflux for 20 hr. The reaction mixture on cooling gave 4f as yellow crystalline solid.

1-Methyl-4-nitro-5-(2-diethylaminoethylamino)imidazole (4g) and 1-methyl-4-nitro-5-[3-[N-piperidino]propylamino]imidazole (4h)—A mixture of 3 (4.85 g), 2-diethylaminoethylamine (7 g) and benzene (60 ml) was heated under reflux for 4 hr, the solvent evaporated off and the residue trituated with water. The solid was filtered off and recrystallised from methylene chloride-ether to afford 4g.

Similarly was prepared 4h from 3 and 3-piperidino-propylamine.

1-Methyl-4-nitro-5-(2-pyrazinyloxy)imidazole (4i), 1-methyl-4-nitro-5-(4-quinazolinylloxy)imidazole (4j) and 1-methyl-4-nitro-5-(3,4,5-trichlorophenoxy)imidazole (4k)—To a solution of 2-hydroxypyrazine (4.8 g) in freshly prepared sodium ethoxide, (1.2 g Na dissolved in 125 ml of absolute ethanol) was added under stirring, a solution of 3 (8.05 g) in hot absolute ethanol

Table 1—1-Methyl-4-nitro-5-substituted imidazoles

Compd.	Yield %	Crystallised from	m.p. (°C)	Mol. formula	Analysis (%)					
					Calculated			Found		
					C	H	N	C	H	N
4a	4	CH <sub>3</sub> CN–Et <sub>2</sub> O	105-9	C <sub>9</sub> H <sub>11</sub> N <sub>5</sub> O <sub>4</sub>	42.69	4.38	27.67	43.06	4.58	27.72
4b	15	CH <sub>3</sub> CN–Et <sub>2</sub> O	136	C <sub>7</sub> H <sub>9</sub> N <sub>5</sub> O <sub>3</sub>	39.81	4.30	33.17	40.20	4.44	32.98
4c	40	CH <sub>3</sub> CN–Et <sub>2</sub> O	134	C <sub>7</sub> H <sub>8</sub> N <sub>4</sub> O <sub>4</sub>	39.62	3.80	26.41	40.22	3.90	27.02
4d	83	CH <sub>3</sub> CN–Et <sub>2</sub> O	233-35	C <sub>7</sub> H <sub>8</sub> N <sub>4</sub> O <sub>3</sub> S	36.85	3.53	24.56	37.12	3.57	24.97
4e	20	H <sub>2</sub> O	160-61	C <sub>7</sub> H <sub>7</sub> N <sub>5</sub> O <sub>2</sub>	43.52	3.65	36.26	43.75	3.83	36.58
4f HCl	78	MeOH–EtOAc	291(d)	C <sub>9</sub> H <sub>9</sub> N <sub>5</sub> O <sub>2</sub> ·HCl	42.28	3.94	27.39	42.05	4.65	27.69
4g	66	CH <sub>2</sub> Cl <sub>2</sub> –Et <sub>2</sub> O	78-82	C <sub>10</sub> H <sub>19</sub> N <sub>5</sub> O <sub>2</sub>	49.77	7.94	29.03	49.41	8.03	29.23
4h	62	CH <sub>2</sub> Cl <sub>2</sub> –Et <sub>2</sub> O	86-88	C <sub>12</sub> H <sub>21</sub> N <sub>5</sub> O <sub>2</sub>	53.92	7.92	26.20	53.72	8.07	25.86
4i	45	DMF–MeOH	> 300	C <sub>8</sub> H <sub>7</sub> N <sub>5</sub> O <sub>3</sub>	43.44	3.19	31.67	43.40	3.28	31.42
4j	45	CHCl <sub>3</sub> –Et <sub>2</sub> O	186-87	C <sub>12</sub> H <sub>9</sub> N <sub>5</sub> O <sub>3</sub>	53.14	3.34	25.82	53.50	3.52	25.72
4k	32	CHCl <sub>3</sub> –hexane	234	C <sub>10</sub> H <sub>6</sub> Cl <sub>3</sub> N <sub>3</sub> O <sub>3</sub>	37.24	1.88	13.03	37.49	1.99	12.89
4l	60	Benzene	134-36	C <sub>10</sub> H <sub>10</sub> N <sub>4</sub> O <sub>2</sub> S	48.00	4.02	22.39	48.02	4.26	22.67
4m	70	EtOAc–hexane	182	C <sub>13</sub> H <sub>13</sub> N <sub>3</sub> O <sub>4</sub> S	50.81	4.26	13.68	51.12	4.37	13.09
4n	72	MeOH	154(d)	C <sub>9</sub> H <sub>8</sub> N <sub>4</sub> O <sub>3</sub> S	42.86	3.20	22.22	42.80	3.33	21.83
5	25	DMF–Et <sub>2</sub> O	286-87(d)	C <sub>11</sub> H <sub>12</sub> N <sub>4</sub> O <sub>5</sub>	39.29	3.60	33.33	39.80	3.88	33.42
6	68	EtOH	166-67	C <sub>12</sub> H <sub>12</sub> N <sub>4</sub> O <sub>3</sub> S	49.31	4.14	19.17	49.54	4.51	19.45

(100 ml) and the reaction mixture boiled under reflux for 18-20 hr. The solvent was evaporated off, the residue treated with water (30 ml) and extracted with methylene chloride. The methylene chloride extract was evaporated *in vacuo* and the residue crystallised from DMF-methanol to afford 4i.

Similarly were synthesised 4j and 4k by the action of 3 on sodium salts of 4-hydroxyquinazoline and 3,4,5-trichloro-phenol respectively.

**1-Methyl-4-nitro-5-(o-aminophenyl)mercaptoimidazole (4l)**—A solution of 3 (4 g) in hot ethanol (50 ml) was added to a solution of o-aminothiophenol (3.13 g) in ethanol (30 ml) containing 12.5% aq. sodium hydroxide (8 ml). The reaction mixture was heated under reflux for 30 min, concentrated *in vacuo*, cooled and the solid obtained filtered off and recrystallised from benzene to afford 4l.

**1-Methyl-4-nitro-5-(2-carbethoxyphenyl)mercaptoimidazole (4m)**—To a solution of ethyl thiosalicylate in sodium ethoxide (2.4 g of Na dissolved in 100 ml of absolute ethanol) was added under stirring a hot solution of 3 (16.15 g) in absolute ethanol (200 ml). The reaction mixture was heated under reflux for 16-18 hr, allowed to cool to room temperature, the solid filtered off, triturated with water and filtered again. Recrystallisation from methanolic chloroform afforded 4m.

**2-(1-Methyl-4-nitroimidazol-5-yl)mercaptopyridine N-oxide (4n)**—A warm solution of 3 (8.05 g) in ethanol (100 ml) was added under stirring to 2-mercaptopyridine N-oxide sodium salt in warm ethanol (80 ml) and the reaction mixture stirred overnight at room temperature. The yellow solid separated was filtered off, triturated with water, filtered again and recrystallised from methanol at  $<50^{\circ}$  (the solution turned dark at higher temperature) to afford 4n.

**1-Methyl-4-nitro-5-(o-acetamidophenyl)mercaptoimidazole (6)**—A mixture of 4l (1 g) and acetic anhydride (5 ml) containing a drop of pyridine was heated on a water-bath at  $90^{\circ}$  for 3 hr. The reaction mixture was cooled, poured into water and left at room temperature for 2-3 hr. The solid was filtered off and recrystallised from ethanol to afford 6.

**1-Methyl-4-amino-5-(2-carbethoxyphenyl)mercaptoimidazole (8)**—A solution of 4m (6.15 g, 20 mmol) in 2-methoxyethanol (150 ml) was hydrogenated in a Parr apparatus at 40 psi at  $45^{\circ}$  in the presence of Raney nickel (2 g) until 60 mmol of hydrogen was absorbed (3 hr). The solution was filtered, the filtrate evaporated *in vacuo* and the residue recrystallised from methylene chloride-hexane to afford 8 (1.6 g), m.p.  $162-63^{\circ}$  (Found: C, 56.6; H, 5.8; N, 15.0.  $C_{13}H_{15}N_3O_2S$  requires C, 56.3; H, 5.5; N, 15.2%).

**1-Methyl-4-(1-imidazolyl)-5-nitroimidazole (11)**—11 was obtained from 10 and imidazole by the method described for 4a. The crude product was converted into the nitrate salt which crystallised from methanol-ether, yield 42%, m.p.  $185-86^{\circ}$  (Found: C, 33.2; H, 3.5; N, 33.0.  $C_7H_7N_5O_2 \cdot HNO_3$  requires C, 32.8; H, 3.2; N, 32.8%).

#### References

- 1 Nagarajan K, Arya V P, George T, Sudarsanam V, Shah R K, Goud A N, Shenoy S J, Honkan V, Kulkarni Y S & Rao M K, *Indian J Chem*, **21B** (1982).
- 2 Merck index (Merck and Co, New Jersey), 1968, 114.
- 3 Sarasin J & Wegman E, *Helv chim Acta*, **7** (1924) 713.
- 4 Ilvespaa A O & Jarumilinta R, *Indian J Chem*, **21B** (1982).
- 5 Galbreath R J & Ingham R K, *J Org Chem*, **23** (1958) 1804; Trace W E, Kreider E M & Brand W W, *Org React*, **18** (1970) 99.
- 6 Nagarajan K, Arya V P, Shah R K, Shenoy S J & Bhat G A, *Indian J Chem*, **21B** (1982).

