# LEATHER LUBRICANTS-PL VI: STUDIES ON OIL TANNING

K. J. KEDLAYA, E. C. MATHEW & M. SANTAPPA

Central Leather Research Institute, Madras.

#### Received on March 29, 1974.

Even though fish oil is the well-known oil tanning agent, other unsaturated oils like rubber seed oil, mustard oil can be used in oil tanning; this indicates that the presence of multiethenoid fatty acid as fatty acid constituent of oil molecule is not a prerequisite for oil tanning.

Oil tanning is one of the oldest modes of converting hides and skins into leather and this is the only tannage wherein the oil incorporated not only tans the skins but lubricates it. Universally known also chamois generally means fish oil-tanned light and loose-textured skins having the grain removed, the name chamois tannage however being derived from primitive tannage of soft skins of goat like antelope Rupicapra tragus. Oil tanning was also done with vegetable oil; since two or three centuries rape seed oil-tanned leather called Japanese white leather was produced by Japan'; these leathers are not called as chamois. This clearly indicates that the tanning ability of oils is not confined to the oils of marine origin-cod, sardine, herring or whale oil. Though fish oil (e.g. sardine oil) and rape seed oil are semi drying oils, their fatty acid constituents are not similar; fish oils contain mono-, di- and poly-ethenoid unsaturated fatty acids while rape seed oil contains mainly mono- and diethenoid unsaturated fatty acids and a very minor amount<sup>2</sup> of triethenoid fatty acids<sup>2</sup> (Table 1). This itself indicated that the presence of unsaturated fatty acid consti-

LEATHER SCIENCE, VOL. 21, 1974.

tuents having more than two double bonds may not be a prerequisite for tanning potency of a given oil; presence of more than two double bonds in fatty acid chains however might contribute to certain products as a result of oxidative changes of oil incorporated into the skins and hence responsible for better physical characteristics of fish oiltanned leathers. The mechanism of tanning by marine oils was well studied and well documented.1.4 The present study is to assess the ability of certain vegetable oils and to assess the role of multi-ethenoid fatty acids of oil molecule in tanning for which purpose oils of seeds of lin, rubber, mustard, tobacco, sunflower and sesame were chosen to tan skins. Skins were also treated with (a) butyl ester of oleic acid and (b) linolcic acid and the ability of these products to tan skins was assessed.

Now-a-days fish oil tannage is mostly carried out as fish oil-formaldehyde combination tannage as the tanning with oil alone requires too much meticulous care in controlling the temperature while stoking and oxidation of fish oil. Hence the possibility of using that vegetable oil/s

TABLE 1

011	lodine	value	Oleic or other C <sub>16</sub> motioene acids	Lin	oleic	Line othe cthe fatt	olenie or er tri+ goid gaeids	C <sub>18</sub> -C <sub>20</sub> monoene	Erucie	Multiethe- noid acids C20-C24	Total un- saturated fatty acids
Cod	140-170										·
Sardine	170-193		13					24	•		
Linseed	17()-204		13	15	·	56		24		45	82
Rubber seed	140		17	36		24			· · ·		84
Mustard sced	106-113	:	31	17		\$	-		47		
Tobacco seed	145	. ;	10	72 <sup>'</sup>		I	•		12.	•	93
Sunflower seed	125-136	2	!3	54							8.3
Sesame	103-116	4	5	41	<u>.</u>		: . '	,			77

IODINE VALUE<sup>®</sup> UNSATURATED FATTY ACIDS OF OILS (PERCENT WEIGHT<sup>\*</sup>)

which proved to possess tanning potency in the above experiment was investigaged.

As the purpose of the present study is not to substitute fish oil in making chamois but to make oil-tanned leathers which may find use for clothing or gloving leather and also to know the effect of the nature of unsaturated fatty acid present in oil tanning, skins with grain-on were chosen as the substrate for the study.

#### Experimental

Delimed pelts (6 nos.) with grain on were treated with 25% solution of an oil in isopropanol. The application of oil in isopropanol facilitates quicker penetration and uniform distribution of oil within the pelt because it reduces viscosity of the oil to be incorporated and also causes the dehydration of the pelt; its ability to be the common solvent for oil and water is another factor in favour of the penetration of oil. The skins thus treated were dried in hot chamber at 40°C for a day and again treated with oil in isopropanol. The process of treating and drying overnight was continued till the oil incorporated was 30% of the pelt weight. After completion of this mode of stuffing, the skins were airdried in hot chamber (40°C) for 10 days; the skins were intermittently staked with hand staking machine for a good opening up of the structure of skins and for good exposure of leather surface to hot air.

The skins are finally degreased with 3.5% of soda ash and 1% of a nonionic wetting agent (Noigen LT 180, Dai-ichi Karkaria-India). For degreasing, the skins were treated first with the solution of half of the specified quantity of detergent in a drum -float-600% of pelt weight, tempera-ture 40-45°C and time 1 hour. The drum was drained, filled with fresh water (600% float) to which half of the specified quantity of soda ash was added, run for 1 hour and

LEATHER SCHENCE, VOL. 21, 1974

drained. This process of washing with soda ash solution was repeated once again using the remaining quantity of soda ash. The skins were then given a final wash using the remaining detergent and then dried; the procedure of final washing is the same as that adopted in the first detergent wash.

The dried skins are staked and buffed first with emery paper of 80 grit and then with 320 grit.

2. Formaldchyde-oil combination tannages

Pickled skins were treated in a drum with 3% of formalin (on pelt weight) for 3 hours after which the skins were depickled by the addition of sodium acetate (3% on pelt weight); as 10% solution was fed to the drum in 4 instalments at an interval of 15 minutes, pH of skins thus depickled was adjusted to 7-8 using sodium bicarbonate Next day, these (1% on pelt weight). leathers were slickered for removing excess water and treated with 30% of oil on peli The oil, before application, was mixed with 8% of calcium carbonate and weight. = treated with  $\frac{1}{2}$ % of soda ash as 5% solution for making its emulsion; the skins were painted with this emulsion uniformly on both sides and drummed for 6 hours for two consecutive days after which period the skins were air-dried in a hot chamber (40-45°C) for 10 days. These leathers were subsequently degreased and further processed as in experiment 1.

Leathers as such and after rewetting overnight and drying were assessed for their colour and\_feel. Static water absorption value of these leathers was determined by Kubelkopf method<sup>7</sup>; wherever necessary rate of sinking in water was determined by I.S.I. "Sink Test" method<sup>8</sup>; free oil content of leathers was determined by Official Method<sup>7</sup> and bound oil by I.S.J. methed<sup>8</sup>.

LEATHER SCIENCE, VOL. 21, 1974.

# Results and discussion

It is seen from Table 2 that all the oils of seeds viz., lin, rubber, mustard, tobacco, sunflower and sesame could tan skins like All these leathers could be repeatedly wetted and dried; their leathery nature is not altered as in the case of fish Among these oiloil-tanned leathers. tanned leathers, sesame oil-tanned leathers were a little empty but on subjecting sesame oil-tanned leathers to tanning with sesame oil once again, the feel of leather Rubber, mustard, tobacco was improved. and sunflower seed oil-tanned leathers were soft and full but not as full as sardine fish oil-tanned leathers; in linseed oil-tanned leather, some amount of resinified oil deposited on the surface at certain places and the leathers were soft at places having less or no deposition. By taking proper care for ensuring complete penetration, the deposition of resinified oil matter on the leather surface could be minimised. This was achieved not only by impregnating the oil as isopropanol solution but also by frequent staking. Rubber, mustard, bacco, sunflower and sesame oil-tanned leathers are fairly white in colour, while (sardine) fish oil-tanned leathers are cream yellow; skins tanned/treated with ester of oleic or linoleic acid were fairly hard. Water absorbing capacity (Table 3) of linseed and sesame off-tanned leathers was much less than that of leathers tanned with sardine fish oil and rubber, mustard, to-Sink tests, bacco or sunflower seed oils. however, showed that fish oil-tanned leather sank in 1 minute while rubber seed oiltanned leather required 10-13 minutes for sinking. Shrinkage temperatures of all these leathers were around 60°C. Formaldehyde + rubber seed or tobacco seed oil-tanned leathers are soft and pale yellowish white in colour. They absorb water well (Table 4).

## TABLE 2

Oil	Characteristics of leather			
1. Sardine fish	Full, soft, deep orange coloured leather. Rewetted leather dries soft.			
2. Linseed	Full, softer than the fish oil-tanned leather & light if there is no deposition of fixed matter; hard in places having resin deposition; leather is light yellow in colour; rewetted leather dries soft if there is no resin deposition.			
. Rubber seed	Less full than fish oil-tanned leather; soft leather, pale yellow in colour; rewetted leather dries soft.			
4. Wustard seed	Pale white leather similar to (3)			
Tobacco seed	Leather similar to (3)			
Sunflower seed	Pale white leather similar to (3)			
. Sesame	Pale white soft leather, less full as compared to (3); other properties are similar.			
Butyl ester of oleic acid	Washed, degreased skin dries joto hard maturist			
Butyl ester of linoleic acid	Washed and degreased skin dries into hard material.			

## CHARACTERISTICS OF VARIOUS OIL-TANNED LEATE

TABLE 3

# WATER ABSORPTION OF OIL-TANNED LEATHER (KUBELKOPF METHOD)

Oil	1 hour	% water absorption 2 hours	24 hours
1. Sardine fish	405	411	427
2. Linseed	210	224	239
A Musturd as 1	408	421	440
5 Tobacco world	373	362	• 432
6. Supflower send	551	587	875
7. Sesame	424	426	451
	217	296	297

CABLE 4

% WATER ABSORPTION OF RUBBER SEED AND TOBACCO SEED OIL-TANNED LEATHER

Leather	% Water absorption					
Farmyddatsy da er men a sa	t hour	2 <sub>s</sub> hours	24 hours			
Formaldehyde 4 rubber send oil	485	516	640			
Formaldehyde 4 tobacco seed oil	426	426 NE	583			
	102	413	498			

LEATHER SCIENCE, VOL. 21, 1974, ...

These results did indicate that the presence of tri-and multi-ethenoid fatty acids i.c., fatty acids having more than three double bonds, is not a prerequisite for oil tanning. Linseed oil, though it tanned the skin, exhibited too much a tendency to resinify and deposit on the surface which is the cause for the decreased water. absorption of linseed oil-tanned leather. This is because linseed oil has more than 50% linoleic acid, a triethenoid fatty acid<sup>2</sup> This fatty acid of linseed oil (Table 1). has too quick a drying tendency and for this reason linseed oil was used in conjunction with fish oil in chamois leather production, particularly, when the oxidative power of the fish oil sample was not upto expectation. The good tanning ability of rubber seed, mustard, tobacco and sunflower seed oils indicates that the presence of a certain minimum amount of di-ethenoid fatty acid (about 17%) (Table 1) is needed, provided the total amount of unsaturated fatty acids is above a certain minimum (about 75%) and the iodine value is 120 and above<sup>2,3</sup>. This is obvious from the percent unsaturated fatty acid composition given in Table 1. The decreased water absorption or emptileathers is ness of sesame oil-tanned perhaps due to the fact that its iodine value is less than 125 and hence its tanning potency is less as compared to other oils studied. The slow sinking of rubber seed oil-tanned leather as compared to fish oiltanned leather even though percent water absorption values of both the leathers were high, indicated the presence of some oil digradation products in rubber seed oiltanned leather. This was further substantiated by the free and bound oil-content of these two leathers; fish oil-tanned leather had 1.5% of free oil and 1.5% of bound oil whereas rubber seed oil-tanned leather had 1.7% of free oil and 3.1% of bound oil (the percentages were on hide substance basis); higher bound oil content of rubber seed

LEATHER SCIENCE, VOL. 21, 1974.

oil-tanned leather indicates that a certain degradation products of the oil are still present in the leather. The fact that butyl esters of oleic or linoleic produce fairly hard leather indicated the possible role of saturated fatty acids present in various oils used Only unsaturated fatty acids in tanning. undergo oxidative changes; saturated fatty acids of glyceride molecule remain in tact. The unsaturated fatty acids undergo oxida-tive degradation and the degradation products involve themselves in tanning; the intact saturated fatty acid of oil molecule containing the unsaturated fatty acid may contribute to lubrication effect. Οn this count, esters like butyl esters of linoleic acid may not produce leather as soft as the one produced by unsaturated oil containing such unsaturated fatty acids.

From the foregoing, it is clear that the tanning potency of a given oil is not related to the presence of multi-ethenoid fatty acids in the oil molecule; since the unsaturated fatty acids like oleic acid present in all the oils chosen for study may produce epoxides' during auto oxidation of oil and since oils like sunflower oil which contain only oleic and linoleic acids as unsaturated fatty acid constituents can tan skin, it is clear that oil tannage, at least partly, is an epoxide tannage ; a semilar view was expressed by others.<sup>4,19</sup>

As the formaldehyde-rubber seed or tobacco seed oil-tanned leathers are soft and supple, they can be used in making gloves and garments; their pale white colour is a point in their favour as compared to orange colour of similarly processed fish oiltanned leathers. The water absorption properties (Kubelkopf) of formaldehyde fish oil (sardine)-tanned leather and formaldehyde-rubber seed or tobacco seed oil-tanned leather (Table 4) are similar i.e., all these leathers are highly water

absorbing in nature. As high water absorbing nature is considered not desirable for garment type of leathers, these skins can be pretanned with chrome and retanned with rubber seed or tobacco seed oil. Preliminary investigations showed that the pretanning of pickled skin with 1%  $Cr_2 0_a$ followed by neutralisation with a neutralising syntan and then retanning with rubber seed or tobacco seed oil resulted in soft well-lubricated chrome-oil combinationtanned leather which may be used in making gloves and garments. It is to be noted that chrome-fish oil cambinationtanned leather made, not by this technique, but by the application of chromic acid and lish oil to the skin wherein fish oil while undergoing oxidation also causes reduction of chromic acid into tanning salts, was recommended for gloves and garments."

It is concluded that semi-drying oils like rubber seed or tobacco seed oil which do not have unsaturated fatty acids containing more than two double bonds can be used in oil tanning and tanning potency of oil is not dependent on the presence of tri--or multi-ethenoid fatty acid in the oil molecule. Formaldehyde or mild chrome-tanned leather  $(1.0\% \text{ Cr}_20_3)$  retanned with rubber or tobacco seed oil may be used in making softer type of leathers like gloves and garments.

#### REFERENCES

1. Sawayama, S., Anniversary number "Osterr. Lederz. Zum. Wienner Kongress für Gerberiechemie und Ledertechnik, Wien, 1954.

4.

2. Hilditch, T. P. & Williams, P. N., "The Chemical constitution of natural fats" Chapman & Hall, London, (1964)

3. Mehlenbacher, V. C., "The analysis of fats and oils" The Garrard Publishers, London (1960)

4. Kuntzel, A., in "The chemistry and technology of leather" Reinhold Publishing Corporation, New York (1958)

5. Jayaraman, K. S., Ph. D. Thesis, University of Madras, (1963)

6. Jayaraman, K. S. & Nayudamma, Y., Leath. Sci., 11, 253 & 358 (1964)

7. Official methods of analysis (1965). Society of Leather Trades Chemists, Redbourn Herts.

8. IS: 1016 15 (1956)

9. IS: 1016 8 (1956)

10. Furmer, F. H., J. Chem. Soc., 121, 139 (1942); 119, 541 (1943) and Trans. Faraday Soc., 38, 340, 348, 356, (1942); 42, 228 (1946)

11. Dohogne, A., Rev. Tech. Ind. Cuir. 55, 220 (1963)

### LEATHER SCIENCE, VOL. 21, 1974.