# Crystal Structure of 2,6-Diisocyano-1,2,3,5,6,7-hexahydro-s-indacene-2,6dicarboxylic Acid Diethylester 

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#### Abstract

The title structure, 2,6-diisocyano-1,2,3,5,6,7-hexahydro-s-indacene-2,6-dicarboxylic acid diethyl ester is an indane-based amino acid derivative. It crystallizes in the tetragonal space group $I 4_{1} / a c d$ with unit cell parameters $a=22.868(1) \AA$, $c=14.385(1) \AA$ and $V=7522.8(1) \AA^{3}$. The residual index of the final refinement is 0.06 for 18734 observed reflections. The five-membered ring is distorted, showing an envelope conformation. The molecular packing is stabilized by C-H...O hydrogen-bonding interactions.


(Received, June 18, 2007; Accepted December 20, 2007; Published on web April 11, 2008)

A peptide backbone conformation or a secondary structure is crucial in the design of peptide-based therapeutics. ${ }^{1}$ In this regard, $\alpha, \alpha$-dialkylated amino acids play an important role in the design of a conformationally restricted peptide. ${ }^{2,3}$ Among that, the cyclic $\alpha$-amino acids are considered to be a special class of $\alpha, \alpha$-dialkylated amino acids. The title molecule (I) is also one among the $\alpha, \alpha$-dialkylated amino acids that has been taken to understand its structure and conformational geometry.
The synthesis details of compound (I) are reported in the literature, ${ }^{4}$ and it was crystallized from a mixture of petroleum ether and ethyl acetate (9:1) solvents. The crystal data and the structure determination details are summarized in Table 1. The structure was solved by direct methods and refined by a fullmatrix least-squares technique. All of the non-hydrogen atoms were refined anisotropically and the H atoms were geometrically fixed and constrained to ride on the parent atom in the model. The atomic coordinates and equivalent isotropic displacement parameters for non-hydrogen atoms are given in Table 2. Selected inter-atomic distances and angles are listed in Table 3.
The asymmetric unit of (I) contains a half of the $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{4}$ unit. This molecule is on the crystallographic center of symmetry, and half of the molecule is independent. The interatomic distances and bond angles of (I) reflect the usual geometry of five and six-membered rings. The C-N distances are unequal $[\mathrm{C}(6) \equiv \mathrm{N}$ is $1.135(4) \AA$ and $\mathrm{C}(2)-\mathrm{N}$ is $1.435(4) \AA]$.


Fig. 1 Chemical structure of the title compound.

[^0]One of them was shortened due to the different hybridization i.e. $s p$ and $s p^{3}$, respectively. Specifically, the $\mathrm{C}(6) \equiv \mathrm{N}$ distance is much shorter than the reported structures $\left[1.157(2) \AA{ }^{\circ}\right]^{6}$ This difference may be attributed to the different environment. The bond lengths involving $\mathrm{C}_{\text {sp } 3}$ atoms range from 1.543(3) to $1.544(4) \AA$, for except $C(8)-C(9)[1.269(6) \AA]$. This difference may be due to the large thermal motion of the $C(8)$ and $C(9)$

Table 1 Crystal data and experimental details

| Molecular Formula | $\mathrm{C}_{20} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{4}$ |
| :--- | :--- |
| Formula weight | 352.38 |
| Temperature | $293(2) \mathrm{K}$ |
| Radiation | $\mathrm{Mo} K_{\alpha}$ |
| Wavelength | $0.71073 \AA$ |
| Crystal system | tetragonal |
| Space group | $I 4_{1} /$ acd $Z=16$ |
| Cell dimensions | $a=22.868(1) \AA$ |
|  | $c=14.385(1) \AA$ |
| Volume | $7522.8(1) \AA^{3}$ |
| Absorption Coefficient | $0.088 \mathrm{~mm}^{-1}$ |
| $\mathrm{D}_{\mathrm{c}}$ | $1.245 \mathrm{Mg} \mathrm{m}^{-3}$ |
| $2 \theta_{\text {max }}$ | $52.7^{\circ}$ |
| Crystal size | $0.3 \times 0.22 \times 0.1 \mathrm{~mm}$ |
| $\mathrm{~F}(000)$ | 2976 |
| $R(F)=0.06$ |  |
| $w R\left(F^{2}\right)$ | 0.162 |
| Goodness-of-fit on $F^{2}$ | 1.044 |
| No. of parameters | 121 |
| $(\Delta / \sigma)_{\text {max }}$ | 0.001 |
| $(\Delta / \rho)_{\text {max }}$ | $0.26 \mathrm{e} \AA^{-3}$ |
| $(\Delta / \rho)_{\text {min }}$ | $-0.18 \mathrm{e} \AA \AA^{-3}$ |
| Measurement | SIEMEN S SMART 1K CCD |
| Program System | Area detector |
| Structure determination | SHELXS97 and SHELXL97 |
| Refinement | Direct methods |
| CCDC | full-matrix |
|  | 663020 |
|  |  |

Table 2 Atomic coordinates ( $\AA \times 10^{4}$ ) and equivalent isotropic displacement parameters of non-hydrogen atoms $\left(\AA^{2} \times 10^{3}\right)$

| Atom | $x$ | $y$ | $z$ | $U_{\text {eq }}$ |
| :--- | :---: | ---: | ---: | ---: |
| $\mathrm{O}(1)$ | $9877(1)$ | $-1136(1)$ | $268(2)$ | $107(1)$ |
| $\mathrm{O}(2)$ | $9632(1)$ | $-984(1)$ | $1725(2)$ | $126(1)$ |
| N | $9060(1)$ | $-410(1)$ | $-371(2)$ | $84(1)$ |
| $\mathrm{C}(1)$ | $8490(1)$ | $-608(1)$ | $1010(2)$ | $68(1)$ |
| $\mathrm{C}(2)$ | $9102(1)$ | $-459(1)$ | $62(2)$ | $68(1)$ |
| $\mathrm{C}(3)$ | $9243(1)$ | $150(1)$ | $1023(2)$ | $76(1)$ |
| $\mathrm{C}(4)$ | $8645(1)$ | $418(1)$ | $1137(2)$ | $62(1)$ |
| $\mathrm{C}(5)$ | $8217(1)$ | $-13(1)$ | $1134(2)$ | $57(1)$ |
| $\mathrm{C}(6)$ | $9015(2)$ | $-354(2)$ | $-1152(3)$ | $128(2)$ |
| $\mathrm{C}(7)$ | $9580(1)$ | $-902(1)$ | $830(2)$ | $78(1)$ |
| $\mathrm{C}(8)$ | $10132(2)$ | $-1346(3)$ | $2040(4)$ | $179(3)$ |
| $\mathrm{C}(9)$ | $10022(3)$ | $-1608(3)$ | $2798(4)$ | $189(3)$ |
| $\mathrm{C}(10)$ | $8500(1)$ | $1000(1)$ | 1250 | $70(1)$ |
| $\mathrm{C}(11)$ | $7631(1)$ | $131(1)$ | 1250 | $59(1)$ |

Table 3 Selected bond distances $(\AA)$ and angles $\left({ }^{\circ}\right)$

| $\mathrm{O}(1)-\mathrm{C}(7)$ | $1.182(3)$ | $\mathrm{N}-\mathrm{C}(2)-\mathrm{C}(1)$ | $108.6(2)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O}(2)-\mathrm{C}(7)$ | $1.307(4)$ | $\mathrm{C}(7)-\mathrm{C}(2)-\mathrm{C}(1)$ | $115.7(2)$ |
| $\mathrm{O}(2)-\mathrm{C}(8)$ | $1.484(4)$ | $\mathrm{N}-\mathrm{C}(2)-\mathrm{C}(3)$ | $108.3(2)$ |
| $\mathrm{N}-\mathrm{C}(6)$ | $1.135(4)$ | $\mathrm{C}(7)-\mathrm{C}(2)-\mathrm{C}(3)$ | $112.1(2)$ |
| $\mathrm{N}-\mathrm{C}(2)$ | $1.435(4)$ | $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $104.6(2)$ |
| $\mathrm{C}(1)-\mathrm{C}(5)$ | $1.509(3)$ | $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(2)$ | $102.6(2)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.543(3)$ | $\mathrm{C}(10)-\mathrm{C}(4)-\mathrm{C}(5)$ | $121.1(2)$ |
| $\mathrm{C}(2)-\mathrm{C}(7)$ | $1.521(4)$ | $\mathrm{C}(10)-\mathrm{C}(4)-\mathrm{C}(3)$ | $128.4(2)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.544(4)$ | $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{C}(3)$ | $110.5(2)$ |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | $1.508(3)$ | $\mathrm{C}(11)-\mathrm{C}(5)-\mathrm{C}(4)$ | $120.7(2)$ |
| $\mathrm{C}(4)-\mathrm{C}(10)$ | $1.381(3)$ | $\mathrm{C}(11)-\mathrm{C}(5)-\mathrm{C}(1)$ | $129.0(2)$ |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.389(3)$ | $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(1)$ | $110.4(2)$ |
| $\mathrm{C}(5)-\mathrm{C}(11)$ | $1.388(3)$ | $\mathrm{O}(1)-\mathrm{C}(7)-\mathrm{O}(2)$ | $123.9(3)$ |
| $\mathrm{C}(8)-\mathrm{C}(9)$ | $1.269(6)$ | $\mathrm{O}(1)-\mathrm{C}(7)-\mathrm{C}(2)$ | $125.4(3)$ |
| $\mathrm{C}(10)-\mathrm{C}(4)^{\mathrm{i}}$ | $1.381(3)$ | $\mathrm{O}(2)-\mathrm{C}(7)-\mathrm{C}(2)$ | $110.7(2)$ |
| $\mathrm{C}(11)-\mathrm{C}(5)^{\mathrm{i}}$ | $1.388(3)$ | $\mathrm{C}(9)-\mathrm{C}(8)-\mathrm{O}(2)$ | $111.9(4)$ |
| $\mathrm{C}(7)-\mathrm{O}(2)-\mathrm{C}(8)$ | $116.7(3)$ | $\mathrm{C}(4)-\mathrm{C}(10)-\mathrm{C}(4)^{\mathrm{i}}$ | $118.4(3)$ |
| $\mathrm{C}(6)-\mathrm{N}-\mathrm{C}(2)$ | $177.6(3)$ | $\mathrm{C}(5)^{\mathrm{i}}-\mathrm{C}(11)-\mathrm{C}(5)$ | $118.2(3)$ |
| $\mathrm{C}(5)-\mathrm{C}(1)-\mathrm{C}(2)$ | $102.7(2)$ |  |  |
| $\mathrm{N}-\mathrm{C}(2)-\mathrm{C}(7)$ | $107.3(2)$ |  |  |
| $x-3 / 4 y+3 / 4$, | $1 / 4)$ |  |  |

(i) $x-3 / 4, y+3 / 4,-z+1 / 4$


Fig. 2 The molecular structure of the title compound, showing 50\% probability displacement ellipsoids and arbitrary spheres for the H atoms.
atoms. The carbonyl bonds fall into three categories: $\mathrm{C}_{\mathrm{sp} 3}-\mathrm{O}$ single bonds [C(8)-O(2): 1.484(4) $\AA$ ], $\mathrm{C}_{\text {sp2 }}-\mathrm{O}$ single bonds [ $\mathrm{C}(7)-$ $\mathrm{O}(2): 1.307(4) \AA$ and $\mathrm{C}=\mathrm{O}$ double bonds $[\mathrm{C}(7)=\mathrm{O}(1)$ : $1.182(3) \AA]$. The values of the torsion angles of the isocyano group and the acetate group at $\mathrm{C}(2)$ and $\mathrm{C}(7)$ have a staggered orientation, since the torsion angle $\mathrm{N}-\mathrm{C}(2)-\mathrm{C}(7)-\mathrm{O}(2)$ is $-178.2(3) ;{ }^{\circ}$ whereas the torsion angle, $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{C}(7)-\mathrm{O}(1)$ $\left[-115.6(3)^{\circ}\right]$, shows a partially eclipsed form at the $\mathrm{C}(2)$ and $\mathrm{C}(7)$ atoms. The torsion angles of $\mathrm{C}(8)-\mathrm{O}(2)-\mathrm{C}(7)-\mathrm{O}(1)$ [6.3(6) ${ }^{\circ}$ ] and $\mathrm{N}-\mathrm{C}(2)-\mathrm{C}(7)-\mathrm{O}(1)\left[3.2(4)^{\circ}\right]$ clear the eclipsed form at that center.
The molecular packing is stabilized by $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogenbonding interactions. Figure 2 depicts the $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interaction between the carbonyl group with the adjacent cyclopentane group of the adjacent molecule. This intermolecular interaction forms a centrosymmetric dimer in the crystal. The C(3)$\mathrm{H}(3) \cdots \mathrm{O}(1)$ hydrogen-bond parameters are $\mathrm{C}(3) \cdots \mathrm{O}(1)$ ii: $3.55(4)$, $\mathrm{H}(3 \mathrm{~B}) \cdots \mathrm{O}(1): 2.59(2) \AA$ and the angle is $168.3(2) .{ }^{\circ}$
[Symmetry code(ii): $-x+2,-y,-z$ ]

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